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Adding the number of protons (12) and the number of neutrons (11) together, reveals the mass of the atom to be 23. Many excited state configurations are possible. A few examples include: 2-7-4, 2-5-6, 1-8-4. They contain different numbers of neutrons.  $(19.99 \times .909) + (20.99 \times 0.03) + (21.99 \times 0.88) + (12.00 \times 9.93) + (13.00 \times 1.07) = 100$

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### Modern Chemistry Chapter 12 Review Answers Liquids And Solids

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## Topic 12 - Acids, Bases and Buffers - A-Level Chemistry

A solution was made by dissolving 12.5 g of  $\text{Na}_3\text{PO}_4$  in 125.00 mL of water. The volume of the resulting solution was 128 mL. Calculate the molarity and molality of the solution.

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$(12.00 \times 9.93) + (13.00 \times 1.07) 100$  Both isotopes contain 6 protons but different numbers of neutrons.  
Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 protons and 7 neutrons. CHEMISTRY ANSWER KEY 27 Na Mg Al 1s22s22p63s23p1 Si 1s22s22p63s23p2 P S 1s22s22p63s23p4

### Quia

12.02 They contain different numbers of neutrons.  $(19.99 \times .909) + (20.99 \times 0.03) + (21.99 \times 0.88) 20$   
 $(12.00 \times 9.93) + (13.00 \times 1.07) 100$  Both isotopes contain 6 protons but different numbers of neutrons.  
Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 Na Mg 1s22s22p63s2

### St. Francis Preparatory School

12 + H<sub>2</sub>O → H<sub>2</sub>O + + 1- The vertical columns of the table should have elements with similar chemical properties. If the elements were arranged by atomic mass, several pairs of elements, such as Ar and K, would be out of position in regards to properties. Both groups follow the octet rule. Group I members achieve an octet by losing an electron,

### Answers to Topic 5 Periodic Table from Review Book

View Test Prep - Chem Key from CHEMISTRY 101 at Taft High School. Chemistry: The Physical Setting Answer Key ANSWERS To Tomc 1 "MW Questions 1. 1 2 3 - 3. 2 " " 5- '3 o. 2 7' 1 3- 3 9.

### Chem Key - Chemistry The Physical Setting Answer Key ...

Review Topics and Practice Evaluation Questions These are the topics a student should know fairly well in order to skip Chemistry 122 and enter directly into Chemistry 123 or 128. The more difficult topics are marked with an asterisk.

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Contrapositive: If two angles are not vertical, then they are not congruent. 132-133 #2, 6, 7, 12-15.  
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